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# MOLECULAR MODELLING OF LANOSTEROL 14α-DEMETHYLASE (CYP51) FROM SACCHAROMYCES CEREVISIAE VIA HOMOLOGY WITH CYP102, A UNIQUE BACTERIAL CYTOCHROME P450 ISOFORM: QUANTITATIVE STRUCTURE–ACTIVITY RELATIONSHIPS (QSARs) WITHIN TWO RELATED SERIES OF ANTIFUNGAL AZOLE DERIVATIVES

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The construction of a three-dimensional molecular model of the fungal form of cytochrome P450 (CYP51) from *Saccharomyces cerevisiae*, based on homology with the haemoprotein domain of CYP102 from *Bacillus megaterium* (a unique bacterial P450 of known crystal structure) is described. It is found that the endogenous substrate, lanosterol, can readily occupy the putative active site of the CYP51 model such that the known mono-oxygenation reaction, leading to  $C_{14}$ -demethylation of lanosterol, is the preferred route of metabolism for this particular substrate. Key amino acid contacts within the CYP51 active site appear to orientate lanosterol for oxidative attack at the  $C_{14}$ -methyl group, and the position of the substrate relative to the haem moiety is consistent with the phenyl-iron complexation studies reported by Tuck *et al.* [*J. Biol. Chem.*, **267**, 13175–13179 (1992)]. Typical azole inhibitors, such as ketoconazole, are able to fit the putative active site of CYP51 by a combination of haem ligation, hydrogen bonding,  $\pi$ - $\pi$  stacking and hydrophobic interactions within the enzyme's haem environment. The mode of action of azole antifungals, as described by the modelling studies, is supported by quantitative structure-activity relationship (QSAR) analyses on two groups of structurally related fungal inhibitors. Moreover, the results of molecular electrostatic isopotential (EIP)



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energy calculations are compatible with the proposed mode of binding between azole antifungal agents and the putative active site of CYP51, although membrane interactions may also have a role in the antifungal activity of azole derivatives.

Keywords: CYP51; Lanosterol; 14\alpha-demethylase; Inhibitors; QSAR; CYP102; Antifungals; Azole derivatives

## INTRODUCTION

The cytochromes P450 (CYP) constitute a superfamily of haem-thiolate enzymes which catalyse a large number and variety of mixed function oxidations (and some reductions) for a vast array of structurally diverse chemicals, both exogenous and endogenous.<sup>1-3</sup> The yeast P450 isoform (CYP51), which mediates the formation of the principal fungal steroid ergosterol by a pathway including C14-demethylation of lanosterol, has been the target for antifungal drug design over many years,<sup>4-14</sup> and it is thought that ergosterol may act in the regulation of yeast cell growth under anaerobic conditions.<sup>9</sup> In addition to the fungal forms, CYP51 isozymes are also present in mammalia and other animal species, where lanosterol  $14\alpha$ -demethylase is involved in *de novo* cholesterol biosynthesis.<sup>9</sup> According to the most recent update on P450 nomenclature,<sup>1</sup> CYP51 proteins have been sequenced from human, rat, wheat and a total of nine different species of fungi, including several yeast species such as: Saccaromyces cerevisiae, Candida albicans and C. tropicalis. Figure 1 presents an alignment between the amino acid sequences of these three yeast forms of CYP51 produced using the GCG package (Genetics Computer Group, Madison, Wisconsin) where there is clear evidence of significant homology ( $\sim 89\%$ ) between the three isoforms in agreement with earlier work.<sup>15</sup> In fact, C. tropicalis has been shown to possess at least seven additional P450s which have been assigned to a separate multigene family, CYP52.<sup>16</sup> Unlike CYP51, however, the CYP52 isoforms are associated with the  $\omega$ -terminal oxidations of alkanes,<sup>10</sup> although the two families are likely to be evolutionarily related despite possessing different substrate specificities.

CYP51 from S. cerevisiae exhibits typical P450 spectroscopic properties<sup>10</sup> and, upon binding lanosterol, produces a Type I spectral change which can be used to determine the enzyme-substrate dissociation constant ( $K_D$ ) of 6.0 µM; and this is in good agreement with the apparent Michaelis constant ( $K_m$ ) of 6.25 µM. The binding affinity for lanosterol to CYP51 can, therefore, be estimated as lying within the range -7.096 to -7.120 kcal·mole<sup>-1</sup> at 25°C,

cyp102 .....T ....MAIVET VID....GIN YFLSLSVTQQ ISILLGVPFV YNLVWQYLYS cvp51cal cyp51ctr ....MAIVDT AID....GIN YFLSLSLTQQ ITILVVFPFI YNIAWQLLYS MSATKSIVGE ALEYVNIGLS HFLALPLAQR ISLIIIIPFI YNIVWQLLYS cyp51sce 20 30 40 10 cyp102 I.KE.MPQP. KTFGELKNLP LLNTDKPVQALMKIADELGEI FKFEAPGRVT LRKDRAPLVF YWIPWFGSAA SY.GQQPYEFFESCRQKYGDV FSFMLLGKIM cyp51cal LRKDRVPMVF YWIPWFGSAA SY.GMQPYEFFEKCRLKYGDV FSFMLLGKVM cyp51ctr LRKDRPPLVF YWIPWVGSAV VY.GMKPYEFFEECQKKYGDI FSFVLLGRVM cyp51sce 70 80 90 60 50 RYLSSORLIKEACDESRF.DK NL.SOALKFV RDFAGDGLFT S.,WTHEKNW cyp102 TVY.LGPKGHEFVFNAKLSDV SAEDAYKHLT TPVFGKGVIY DCPNSRLMEQ cyp51cal TVY.LGPKGHEFIYNAKLSDV SAEEAYTHLT TPVFGKGVIY DCPNSRLMEQ cvp51ctr TVY.LGPKGHEFVFNAKLADV SAEAAYAHLT TPVFGKGVIY DCPNSRLMEQ cyp51sce 100 110 120 130 140 KKAHNILLPSFSQQAMKGYHAMMVDIAVQLVQKWERLNADEHI EV. PEDMTRL cyp102 cyp51cal KKFAKFAL.T.TD.SFKRYVPKIREEILNYFVTDESFKLKEKT HGVANVMKTQ cyp51ctr KKFAKFAL.T.TD.SFKTYVPKIREEVLNYFVNDVSFKTKERD HGVASVMKTQ KKFVKGAL.T.KE.AFKSYVPLIAEEVYKYFRDSKNFRLNERT TGTIDVMVTQ cyp51sce 150 160 170 180 190 200 TLDTIGLCGFNYRFNSFYRDQPHPFITSMVRALDEAMNKLQRA.NPDDP.AYDE cyp102 PEITIFTASRSL.F....GDEMRRIFDRSFAQLYSDLDKGFTPINFVFPNLPLP cyp51cal cyp51ctr PEITIFTASRCL.F....GDEMRKSFDRSFAQLYADLDKGFTPINFVFPNLPLP PEMTIFTASRSL.L....GKEMRAKLDTDFAYLYSDLDKGFTPINFVFPNLPLE cyp51sce 210 220 230 240 NKRQFQEDIKVMNDLVDKIIA DRKASGE.Q. SDDLLTHMLN GKDPETGEPL cyp102 HYWRRDAAQKKISATYMKEIK LRRERGDIDP NRDLIDSLLI HSTYKDGVKM cyp51cal HYWRRDAAORKISAHYMKEIK RRRESGDIDP KRDLIDSLLV NSTYKDGVKM cyp51ctr cyp51sce HYRKRDHAQKAISGTYMSLIK ERRKNNDIQ. DRDLIDSLMK NSTYKDGVKM 250 260 270 280 290 DDENIRYQII TFLIAGHETT SGLLSFALYF LVKNPHVLQK AAEEAARVLV cyp102 TDOEIANLLI GILMGGOHTS ASTSAWFLLH LGEKPHLODV IYQEVVELLK cyp51cal TDQEIANLLI GVLMGGQHTS ASTSAWFLLH LAEQPQLQDD LYEELTNLLK cyp51ctr cyp51sce TDQEIANLLI GVLMGGQHTS AATSAWILLH LAERPDVQQE LYEEQMRVL. 300 310 320 330 340 D.P....VPS YKQVKQLKYV GMVLNEALRL W.PTAPAFSL YAKEDTVLGG cyp102 EKGGDLNDLT YEDLQKLPSV NNTIKETLRM HMPLHSIFRK VTNPLRIPET cvp51cal EKGGDLNDLT YEDLQKLPLV NNTIKETLRM HMPLHSIFRK VMNPLRVPNT cvp51ctr cyp51sce D.GG.KKELT YDLLQEMPLL NQTIKETLRM HHPLHSLFRK VMKDMHVPNT 350 360 370 380 FYPLEKGDEL MVLIPQLHRD KTIWGDDVEEFRPERFE.... NPSAI..... cyp102 NYIVPKGHYV LVSPGYAHTS ERYF.DNPEDFDPTRWDTAAA KANSVSFNSS cyp51cal cyp51ctr KYVIPKGHYV LVSAGYAHTS DRWF.EHPEHFNPRRWESDDT KASAVSFNSE cyp51sce SYVIPAGYHV LVSPGYTHLR DEYF.PNAHQFNIHRWN.... KDSASSYSVG 390 400 410 420 cyp102 ... PQHAFKP FGNGQRACIG QQFALHEATL VLGMMLKHFD cyp51cal DEVDYGFGKV SKGVSSPYLP FGGGRHRCIG EQFAYVQLGT ILTTFVYNLR DTVDYGFGKI SKGVSSPYLP FGGGRHRCIG EQFAYVQLGT ILTTYIYNFK cyp51ctr cyp51sce EEVDYGFGAI SKGVSSPYLP FGGGRHRCIG EHFAYCQLGV LMSIFIRTLK 450 430 440 cyp102 FE..DHTNY. ELDIKETLTLKPEGFVVKAKS KKIPLGGI WTI.DGYKVP DPDY.SSMVVLPTEPAEIIWE KRETCMF. cvp51cal cyp51ctr WRL.NGDKVP DVDY.QSMVTLPLEPAEIVWE KRDTCMV. cyp51sce WHYPEGKTVP PPDF.TSMVTLPTGPAKIIWE KRNPEQKI

FIGURE 1 An alignment between CYP102 haemoprotein domain sequence and those of three CYP51 proteins from *S. cerevisiae*, *C. albicans* and *C. tropicalis*, respectively using information from Ref. [1].

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using the expression:

$$\Delta G = RT \ln K$$

where R is the gas constant, T is the absolute temperature, and K may be either  $K_{\rm D}$  or  $K_m$  in this instance.

Based on sequence comparisons, the CYP51 proteins from S. cerevisiae and C. tropicalis exhibit a 66.5% identity and an overall homology of 89.6% when conservatively replaced amino acids are taken into account.<sup>17</sup> It is possible to identify four highly homologous regions in the CYP51 sequences which appear to correspond with the putative active site of the enzyme, including the haem-binding domain. In fact there is a close correspondence between conserved regions in the CYP51 alignment (Figure 1) and the substrate recognition sites (SRSs) identified by Gotoh<sup>18</sup> in a multiple sequence alignment of several CYP2 family proteins. Site-directed mutagenesis of glycine-310 to aspartate (i.e., the change G310D) in CYP51 from S. cerevisiae leads to a catalytically inactive form of the enzyme.<sup>19</sup> Sequence alignment with P450<sub>cam</sub> (CYP101), which was the only P450 available at the time to have been characterized crystallographically, indicated that position 310 in CYP51 lies within the putative I helix which is orientated distal to the haem moiety.<sup>19</sup> Moreover, molecular modelling of part of this distal helix suggested that a rotation of this segment of secondary structure could occur in the mutant form of the enzyme, such that a downstream histidine residue (His-317) would readily ligate the haem iron,<sup>19</sup> thus providing an explanation for the inactivity of the G310D mutant.

It is possible that histidine-317 may play a key role in the action of azole antifungal agents, as it appears that the more potent CYP51 inhibitors contain either an imidazole or triazole 5-membered ring $^{4,11-13}$  which is analogous to the side-chain of a histidine residue. Crystallographic studies on P450<sub>cam</sub> containing a bound imidazole antifungal compound<sup>20</sup> indicate that conformational changes in the enzyme take place upon inhibition, although these appear to affect active site amino acid side-chains rather than the protein backbone itself. However, it is clear from this and other reported studies on imidazole inhibitors (reviewed in Ref. [20]) that the imidazole ring sp<sup>2</sup> nitrogen ligates the haem of P450 at a distance of about 2Å from the iron atom. Consequently, His-317 in CYP51 could be involved in the access of azole inhibitors to the active site, where its normal endogenous function is likely to be associated with the oxygenation mechanism, which requires that the substrate remains enzyme-bound during the entire reaction sequence. involving three successive mono-oxygenations without the release of intermediates.9.10

Morris and Richards<sup>21</sup> reported the results of active site modelling for CYP51 based on homology with CYP101, although sequence alignments using the CYP101 template indicate several regions of poor homology and large gaps due to stretches of unmatched peptide between the two protein sequence<sup>22</sup> which hamper the production of a satisfactory model. However, since the publication of the three-dimensional structure of the CYP102 haemoprotein domain<sup>23</sup> it has been apparent that this represents an improved template for the construction of microsomal P450s<sup>2</sup> since the above problems are largely eliminated. Consequently, homology modelling of CYP51 from the unique bacterial P450 CYP102 crystal structure is likely to advance our understanding of the yeast lanosterol C-14 demethylase system.

# **METHODS**

Figure 1 shows an alignment between three CYP51 sequences from C. tropicalis, C. albicans and S. cerevisiae with that of the CYP102 haemoprotein domain. This alignment was produced using the GCG package (Genetics Computer Group, Maidson, Wisconsin) from the primary structures of the relevant proteins obtained from the most recent update of P450 accession numbers<sup>1</sup> including that of S. cerevisiae.<sup>24</sup> Apart from an N-terminal membrane-binding peptide of about 40 residues, there is a close match between CYP102 and the three CYP51 sequences with only a small number of short gaps, sometimes involving only one or two amino acid residues, with the exception of an 18-residue stretch upstream of the P450 signature motif containing the invariant cysteine which is the proximal haem ligand. As this octadecapeptide does not feature in the CYP52 sequences<sup>16</sup> it would appear that it is a characteristic of CYP51, and may relate to the mode of membrane binding and/or topology of the enzyme's orientation within the phospholipid bilayer, which could have a bearing on catalytic activity. Nevertheless, there is clear homology between CYP102 and CYP51 which supports the use of the former as a structural template and, moreover, we have shown<sup>25</sup> previously that the CYP102 crystal structure is satisfactory for homology modelling of other steroidogenic P450s including aromatase (CYP19) which exhibits significant sequence similarity<sup>1</sup> with CYP51, together with an analogous mechanism of substrate oxygenation.<sup>9,10</sup> Furthermore, CYP52A1, which catalyses the  $\omega$ -hydroxylation of lauric acid, appears to cluster with CYP102 in a dendrogram of eukaryotic P450s<sup>1</sup> where the substrate specificity of CYP102 for long chain fatty acids may be related to the similarities in primary sequences of the two proteins.

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A three-dimensional model of CYP51 was constructed from the CYP102 structure, based on the alignment shown in Figure 1, using the Biopolymer module of the Sybyl package version 6.4 (Tripos Associates, St. Louis, Missouri) for residue replacement, deletion and insertion as required by the sequence alignment. Molecular modelling via Sybyl version 6.4 was carried out on a Silicon Graphics Indigo<sup>2</sup> IMPACT 10000 workstation. All residue insertions were incorporated via loop-searching of the Brookhaven Protein Databank irrespective of the length of peptide required, and the raw structure of CYP51 was then energy minimized using the Tripos force field to produce a low energy minimum geometry of  $-1508.9 \,\mathrm{kcal}\,\mathrm{mole}^{-1}$  in the absence of substrate. The final CYP51 model was then probed using the endogenous substrate, lanosterol, and azole inhibitors, such as ketoconazole and itraconazole, using either crystal structures of the compounds themselves<sup>26-28</sup> or pre-minimized structural fragments obtained from the Sybyl Fragment Library. Crystallographic data for itraconazole was kindly supplied by Henri Moereels (Janssen Pharmaceutica, Beerse, Belgium) from the original crystal coordinates produced by Professor C.J. De Ranter. In both ketoconazole and itraconazole, the active enantiomer was used, namely, the RS form. Interactive docking of substrate and inhibitors within the putative active site of CYP51 was carried out using the DOCK routine in Sybyl 6.4, and involved a 3Å distance constraint between the C-14 methyl and haem iron in the case of lanosterol whereas a 2Å distance constraint between ligating nitrogen and iron was employed for the azole inhibitors, in keeping with crystallographic data on P450 complexes. Quantitative structureactivity relationships (QSARs) for two series of azole antifungals (see for Figure 2 structures) were generated using the reported biological data on CYP51 inhibition<sup>11,29</sup> and from CNDO/2-calculated electronic structures<sup>30</sup> via the COSMIC modelling framework<sup>31</sup> as implemented on a MicroVAX II mini computer operating under VMS. The COSMIC software package was a generous gift from Dr J.G. Vinter.

### **RESULTS AND DISCUSSION**

#### 1. CYP51 Homology Model

Figure 3 shows a view of the energy minimized structure of CYP51 where the lanosterol molecule can be visualized as occupying an essentially hydrophobic pocket in the vicinity of the putative active site bounded by the haem moiety, part of the distal I helix, the F and B' helices, and two  $\beta$ -sheet regions corresponding to alignment positions 328 and 437





FIGURE 2 Structures of all compounds studied in this work together with the numbering system of the azole ring.

(CYP102 nomenclature). Tuck *et al.*<sup>32</sup> have shown that lanosterol is likely to become orientated relative to the haem of CYP51 such that its 3-hydroxyl group can form a hydrogen bond with an active site residue and, in the current model, it appears that serine-382 (at alignment position 329)



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FIGURE 3 A view of the putative active site of CYP51 showing the disposition of the endogenous substrate, lanosterol, orientated for  $14\alpha$ -demethylation. See Color Plate I.

could fulfil the role of a hydrogen bond-forming side-chain which is able to donate a hydrogen bond to lanosterol in CYP51 (Figure 2). Moreover, a second hydrogen bond donor/acceptor (as required by the currently accepted mechanism which is described in Ref. [9]) may be afforded by histidine-381 (at position 328 in the alignment), although histidine-317 (corresponding to alignment position 267) could also fulfil this requirement. The likelihood of a hydrogen bond-forming amino acid side-chain at about this region of the CYP51 active is suggested by the finding that the enzyme exhibits higher affinity for both C-14 hydroxymethyl lansterol and the corresponding aldehyde than for lanosterol itself.<sup>32</sup>

In addition to these hydrogen bond contacts, it is clear from Figure 3 that there are several hydrophobic interaction points between substrate and enzyme, including some evidence of  $\pi-\pi$  stacking interactions. In particular, the planar dimethyl alkene function terminating the D-ring side-chain grouping of lanosterol is able to  $\pi$  stack with the aromatic ring of phenylalanine-153 (position 99 in the alignment). Also, a number of aliphatic amino acid side-chains are implicated in the binding of lanosterol via complementary contacts with substrate methyl groups, in addition to hydrocarbon segments of the steroid nucleus; these hydrophobic amino acid residues include: Val-138, Ile-139, Leu-228, Leu-257, Met-313, Met-509 and Val-510. Furthermore, the aromatic ring of Tyr-140 could also form a  $\pi$ - $\pi$  stacking interaction with a planar unsaturated region of lanosterol (see Figure 3). Consequently, the putative active site of CYP51 represents a closely complementary surface for the endogenous substrate, lanosterol, to bind with high affinity and in such a manner that the molecule is orientated by the CYP51 protein for C<sub>14</sub>-demethylation and with the required stereospecificity.

When one considers a typical CYP51 inhibitor, such as ketoconazole, for example, the situation is fairly analogous to that evidenced with lanosterol binding to CYP51 described above. Figure 4 shows a view of ketoconazole interacting within the putative active site of CYP51 such that inhibition of the enzyme would be brought about via imidazole ligation of the haem iron. To some extent, the same type of amino acid residues are involved in the binding of ketoconazole to the active site of CYP51 as described previously for lanosterol. In particular, the side-chains of Phe-134 and Tyr-126 are able to form  $\pi$ -stacking interactions with the two aromatic rings on the inhibitor, as shown in Figure 4. Furthermore, histidine-381 donates a hydrogen bond to the phenoxy oxygen atom on the ketoconazole side-chain. These, together



FIGURE 4 The inhibitor, ketoconazole, is shown docked within the putative active site of CYP51 where complementary interactions orientate the molecule for haem ligation. See Color Plate II.



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with a number of complementary hydrophobic interactions, orientate the ketoconazole molecule relative to the haem moiety such that the imidazole group lies directly above the central iron atom. In such a position, as shown in Figure 4, the sp<sup>2</sup> nitrogen on imidazole is relatively close to the haem iron at a distance of 3.458 Å. Since many azole antifungal agents possess a 2,4-dichlorophenyl group at an analogous location relative to their imidazole (or triazole) ring as that of ketoconazole, it is likely that these compounds will also occupy the CYP51 active site in a similar manner. Furthermore, the variations in their antifungal activity can be rationalized by the extent of favourable contacts made by the remainder of the molecule, which also extends into the binding site channel like ketoconazole.

# 2. Quantitative Structure-Activity Relationships (QSARs) in Antifungals

Several comparative studies have been reported for azole antifungal agents<sup>11,29,33</sup> including some investigations in mammalian systems.<sup>34–38</sup> It is generally accepted that the evidence from Type II binding spectra indicates azole ligation of the haem moiety in P450 as the mechanism of inhibition<sup>39,40</sup> in contrast to the Type I binding exhibited by the substrate, lanosterol. For the generation of QSARs between antifungal activity and structural descriptors on the compounds concerned, we have used the biological data produced by Pye and Marriott<sup>29</sup> and Vanden Bossche *et al.*<sup>11</sup> as these comprise inhibitory information on a sufficient number of structurally related azoles to permit QSAR analysis.

Table I presents the relevant data set for a series of azoles exhibiting antifungal activity towards C.  $albicans^{29}$  where a number of structural descriptors from a total of 40 independent variables gave rise to statistically significant correlations with both Minimum Inhibitory Concentration (MIC) for arresting fungal growth and inhibition of ergosterol production  $(IC_{50})$ . As far as the former activity is concerned, the only single descriptor correlation (R = 0.77) was that involving  $Q_3$ , which is the net atomic charge on atom 3 of the azole nucleus and adjacent to the haem-ligating nitrogen (see Figure 2 for nomenclature). The value of  $Q_3$  does not vary dramatically within the series, apart from in the triazole compound, butaconazole, which is the least potent antifungal in the series. However, elimination of the data for tioconazole from the analysis results in a substantially improved correlation (R = 0.97) between Q<sub>3</sub> and log MIC, as shown in Table I (Eq. (1)). Plotting out the relationship reveals, however, that it consists of a cluster of five points such that the strength of the regression is primarily determined by the remaining point, corresponding to the triazole butaconazole. Consequently,



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	Compound	$\varrho_3$	$Q_2H$	$lpha\cdot\Delta E$	$\log D_{7.4}$	log MIC	pIC <sub>50</sub>
1	Ketoconazole	0.127	0.0001	678.97	3.30	1.6628	1.6900
7	Miconazole	0.121	0.0610	480.30	6.26	1.4771	2.3010
3	Clotrimazole	0.127	0.0416	474.51	5.68	1.5185	2.0792
4	Econazole	0.122	0.0525	471.17	5.48	1.5185	1.9542
5	Parconazole	0.126	0.0602	474.44	3.19	1.5315	1.7782
9	Butaconazole	0.184	0.0920	475.50	3.34	2.0792	1.7782
۲	Tioconazole	0.120	0.0299	437.67	5.26	0.9243	1.6990
	Regression equation	и	S	R	F		
(I)	$\log MIC = 9.12 Q_3 + 0.40$ (+106)	9	0.058	0.97	73.7		
(2)	$\log MIC = 13.98 Q_2 H + 0.045 \alpha \Delta E - 1.40$ (±2.51) (±0.001)	7	0.134	0.95	17.3		
(3)	$pIC_{50} = 0.15 \log D_{7,4} + 1.24$ (±0.03)	9	060.0	0.94	28.2		
(4)	$pIC_{50} = 3.14 Q_1 H - 5648.95 Q_2 L + 1.76$ (±0.57) (±1034.74)	7	0.088	0.95	17.8		

TABLE I Electronic structural and physicochemical data for 7 azole antifungals v. C. albicans

 $Q_3 =$ met atomic charge on azole ring carbon atom 3;  $Q_1H, Q_2H =$  HOMO electron density on azole ring atoms 1 and 2, respectively ( $\Delta E$ ) where HOMO is the highest occupied MO;  $Q_1H, Q_2H =$  HOMO electron polarizability ( $\alpha$ ) and excitation energy ( $\Delta E$ ) where  $\Delta E = E_{LUMO} - E_{HOMO}$ ;  $\log D_{7,4} =$  common logarithm of the distribution coefficient at pH 7.4;  $Q_2L =$  LUMO electron population on azole ring incogen atom 3 where LUMO is the lowest unoccupied MO;  $Q_2L =$  LUMO electron population or coefficient at pH 7.4;  $Q_2L =$  LUMO electron population concentration (M) against *C. albicans* C66 growth:<sup>39</sup>  $Q_2L =$  -log concentration (M) required for 50% inhibition of ergosterol synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 50% inhibition of ergosterol synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition of ergosterol synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for 10% inhibition synthesis in *C. albicans*.<sup>29</sup>  $Pl_{C_3} =$  -log concentration (M) required for the data for the concentration for the azole grouping.

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this apparently important relationship has to be regarded with considerable suspicion.

Fortunately, the regression analysis also generated another statistically significant correlation which involved a combination of two independent variables,  $Q_2H$  and  $\alpha \cdot \Delta E$  that are, respectively, the HOMO electron population density on  $Q_2$  and the product of polarizability ( $\alpha$ ) and  $\Delta E$ , which is the difference between the energies of the lowest unoccupied and highest occupied molecular orbitals, respectively. As the  $Q_2H$  parameter represents a measure of the electron-donating capacity of the azole nitrogen, which is probably involved in haem ligation of the yeast P450, the appearance of this structural descriptor in a QSAR governing inhibition of fungal growth is to be expected. However, the rationale for the combined parameter  $\alpha \cdot \Delta E$ requires a little more explanation. Essentially, the product of molar polarizability ( $\alpha$ ) and the excitation energy ( $\Delta E$ ) represents an approximation for the dispersion forces (sometimes referred to as London forces or van der Waals forces) of attraction experienced by one molecule interacting with another (reviewed in Ref. [41]). This contribution to the overall binding energy is highly distance-dependent, being inversely proportional to the sixth power of the distance between interacting species, and relates to the number and distribution of electrons over the whole molecule.

This so-called polarization term also represents a component of the solvation energy and, consequently, the partitioning energy between two solvents experienced by a solute molecule,<sup>42</sup> such that it can be regarded as one of the three main factors contributing to the partition coefficient of a compound between, for example, octanol and water. Consequently, the lipophilicity parameter (log  $P_{oct}$ ), for a particular chemical, contains a polarizability component which can be the major factor of its overall lipophilicity.<sup>43,44</sup> One might anticipate, therefore that polarizability will correlate with  $\log P_{\rm oct}$  but it is found that this only occurs for neutral non-polar molecules.<sup>43</sup> Returning to the QSAR expression Table I, Eq. (2), it would appear that the antifungal potency of the seven azoles is governed by the relative ability of the azole nitrogen to ligate the haem group of the yeast P450 isoform partially involved in fungal growth, together with the extent of binding between the inhibitor molecule and the relevant P450 protein (or membrane phospholipid) as determined by the polarization component of compound lipophilicity.

The possibility of compound lipophilicity being relevant to the antifungal activity of the azoles is reinforced by the finding that inhibition of CYP51 for six compounds is proportional to  $\log D_{7,4}$ , the ionization-corrected  $\log P$  value, as shown in Table I, Eq. (3). However, as in the case of the QSAR

relating to MIC Table I, Eq. (1) the data for the thiophene compound, tioconazole, have been excluded from the analysis although the correlation is still acceptable (R = 0.76) for all seven antifungals. The log distribution coefficient, log  $D_{7.4}$ , represents the compound lipophilicity (i.e. log  $P_{oct}$ ) corrected for ionization at pH 7.4<sup>45</sup> and, consequently, contains a component relating to the chemical's basicity due to the presence of the azole ring, together with being a measure of the overall hydrophobic character of the molecule. The good correlation (R = 0.94) with this parameter which, unlike  $Q_3$ , exhibits a satisfactory spread of values within the series of 7 compounds, provides a degree of confidence in the significance of this result which, moreover, can be readily rationalized in terms of the likely mode of binding between the azole antifungals and the CYP51 binding site where hydrophobic interactions appear to feature prominently, in addition to the haem binding ability of the azole ring.

However, the only significant two-parameter correlation obtained for all 7 compounds in the data set involves a combination between two frontier orbital population densities on the azole ring, as shown by Eq. (4), Table I. This expression bears some resemblance to Eq. (2), in that a HOMO electron density is involved and the azole nitrogen likely to be involved in haem iron ligation also features. However, in this case, it is the LUMO density on the nitrogen at position 2  $(Q_2L)$  which is paired with the HOMO electron population on the adjacent carbon  $(Q_1H)$  to produce a satisfactory correlation (R = 0.95) with CYP51 inhibition. Both of these expressions, which are shown as Eqs. (3) and (4) in Table I, exhibit close agreement between calculated and experimental values for  $pIC_{50}$ . Here econazole represents the only slight outlier in each case in that, with Eq. (3) it gives an overestimate whereas Eq. (4) produces an underestimate of econazole's activity. Nevertheless, it is important to find that structural descriptors for this series provide fairly satisfactory correlations with antifungal potency, as the latter depends also on other interactions, including those involving yeast cell membranes both pericytoplasmic and organelle-located.

For the second series of seven azoles, however, some similarities and differences are apparent in the QSARs produced against *S. cerevisiae* inhibition, as presented in Table II, which also lists the relevant parameters involved. Although the biological data stem from an independent study to the one described previously, two compounds (ketoconazole and miconazole) are common to both investigations, although these antifungals appear to be somewhat less potent than, for example, itraconazole.<sup>4</sup> Of the two correlations listed in Table II, Eq. (1) appears to be slightly more significant (Figure 5 compares experimental and calculated values) when one compares



	TABLE II Elec	tronic and moleci	ılar structural	data for 7 az	cole antifungals	v. S. cerevisiae		
	Compound	$Q_4$	$E_{\min}$	$a/d^2$	Ειυμο	pIC <sub>50</sub>	pICsolc.(1)	pICcalc.(2)
-	Itraconazole	-0.018	-12.8	2.3	1.8939	1.6576	1.6367	1.5976
7	Terconazole	-0.025	-11.8	2.2	2.0572	1.4949	1.3930	1.4168
e	Parconazole	-0.015	-18.5	2.3	2.0572	1.4685	1.4511	1.5235
4	Propiconazole	-0.019	-17.0	2.1	1.9592	1.3979	1.4461	1.4469
ŝ	Ketoconazole	-0.018	-26.4	2.0	1.9783	1.2596	1.3662	1.2065
9	Miconazole	-0.028	-15.3	1.3	1.8994	1.0969	1.0486	1.2263
7	Imazalil	-0.033	-17.5	1.8	2.1198	1.0555	1.0891	1.0134
	Regression equation	и	S	R	F			
(1)	$pIC_{s0} = 0.58a/d^2 - 1.14 E_{LUMO} + 2.45$	٢	0.084	0.95	18.6			
(2)	$pIC_{s0} = 29.94Q_4 + 0.03E_{min} + 2.50$ (±0.10) (±0.41)	٢	0.095	0.94	14.0			
pIC 40 =	-log concentration required for 50% inhibition of	S. cerevisiae growth	11					

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 $Q_4$  = net at  $O_{12}$  is the state on introgen atom 4 in the azole ring;  $E_{min} = \text{electrostatic potential energy minimum (kcal.mole<sup>-1</sup>);}$   $E_{L_{1}MO} = \text{energy of the lowest unoccupied molecular orbital (eV.);}$  a/d = ratio of molecular area to the square of molecular depth.



FIGURE 5 Comparison between observed and calculated  $pIC_{50}$  values for 7 compounds based on the data presented in Table II, Eq. (1).

the statistics, although comparisons between the experimental pIC<sub>50</sub> values and those calculated from the two QSAR expressions show that Eq. (2) (Table II) gives a better fit for terconazole, propiconazole and ketoconazole. The two equations in (Table II) contain quite different pairs of structural descriptors: Eq. (1) represents a combination of molecular planarity (area/ depth<sup>2</sup>) and the LUMO energy value ( $E_{LUMO}$ ), whereas Eq. (2) involves the net atomic charge ( $Q_4$ ) on the substituted nitrogen of the azole ring and the energy of the electrostatic potential minimum ( $E_{min}$ ) which lies in the vicinity of the azole nitrogen which ligates the haem iron in CYP51.

However, both equations appear to suggest that there may be interactions between the azole ring and the haem of the fungal P450, which could be synergistic in nature in that there is a possibility of both electron donation from the azole ring to the iron and electron acceptance by the azole system



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brought about by back-donation from the haem moiety. Consequently, the QSAR analyses performed on the two series of antifungals indicate that there are strong reinforcing interactions between the azole ring and the CYP51 haem moiety which help to explain the potency differences in the various compounds. In conclusion, therefore, we feel that the results of QSAR analysis are consistent with the three-dimensional modelling studies of CYP51, because active site interactions help to explain some of the variations in antifungal activity shown by the compounds investigated in yeasts. It is hoped that these parallel studies will assist in the rationalization and prediction of antifungal potency for other structurally related chemicals in both clinical and industrial applications. Future utilization of related designer chemicals may also prove to be valuable as antifungals in both soil and aquatic environments.

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